# Tracing the fate of nutrients in agricultural catchments by stable isotope techniques



## Part I: The Hessian Ried Story

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#### **Outline**

#### Background

- The nitrate problem in Europe
- The monitoring/analytical/experimental approach

#### Results

- Part I: The Hessian Ried Story:
  - Denitrification in groundwater
- Part II: The Selke Story:

Denitrification hot spot in a river stretch

Part III: The Bode Story:

Denitrification potential in a mesoscale river catchment

Conclusions





#### Nitrogen surplus in Europe in kg/ha (2005)



Figure 4: Estimated nitrogen surplus (the difference between inputs and uptake by crops, meat or milk production) in 2005 for Europe. Surplus nitrogen in the soil as a result of excessive application rates and/or low plant uptake can cause an increase in the mineralisation of organic carbon, which in turn leads to an increased depletion of carbon from soils.

Source: JRC/Bouraoui et al., 2009.

# Nitrogen surplus and nitrate in groundwater

What are the resulting nitrate concentrations in leachate water considering nitrogen surplus?



# Nitrate in groundwater - Germany

Status of groundwater bodies with respect to nitrate



#### Situation

About 27% of all GW bodies in Germany are in a poor chemical state due to high nitrate concentrations.

Distribution of nitrate concentrations in network monitoring wells



### Potential solutions for the nitrate problem

Sustainable solution:

Reduction of N-surplus by good agricultural practice.



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#### Sustainable solution:

Reduction of N-surplus by good agricultural practice.

#### Non-sustainable solution:

Natural attenuation concept

#### Natural Attenuation

"Variety of physical, chemical, or biological processes that are at work in-situ, under favourable conditions, acting without human intervention to reduce the mass, toxicity, mobility, volume or concentration of contaminants" (NRC 2000)

Main processes "Nitrate Attenuation":

- Assimilation and conversion to biomass (soil)
- Microbial nitrate reduction (denitrification)
- Dilution (mixing, diffusion, dispersion)
- > Sorption

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### Potential solutions for the nitrate problem

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### **Denitrification - Reactions:**

 $4 \text{ NO}_3^- + 5 \text{ CH}_2\text{O} \rightarrow 2 \text{ N}_2 + 4 \text{ HCO}_3^- + \text{CO}_2$ 

 $14 \text{ NO}_3^- + 5 \text{ FeS}_2 + 4 \text{ H}^+ \rightarrow 7 \text{ N}_2 + 10 \text{ SO}_4^{2-} + 5 \text{ Fe}^{2+}$ 



# The field site: Hessian Ried



Alluvial plain in the upper Rhine valley between the rivers Main (north), Rhine (west), Neckar (south), and the Odenwald mountains (east).

Thick sedimentary fill of the graben structure  $\rightarrow$  great aquifer.

Covers an area of approx. 1200 km<sup>2</sup>.

33% of the area is intensively used for agriculture.

Serves as a major drinking water source for the Frankfurt metropolitan area

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# The experimental/monitoring approach

- Estimation of the denitrification potential using sediments from drilling cores
  - Sulfide- / Disulfide sulphur (CRS- Method, extern)
  - Organic carbon (Liquitoc II, TUD)





# The experimental/monitoring approach

- Estimation of the denitrification potential using sediments from drilling cores
  - Sulfide- / Disulfide sulphur (CRS- Method, extern)
  - Organic carbon (Liquitoc II, TUD)
- Determination of prevailing denitrification pathways
  - Batch experiments
  - Column experiments





## Nitrate reduction potential in drilling cores





# Column experiments

# Isotope effects during denitrification

enrichment of heavy N and O isotopes in residual nitrate

(Rayleigh fractionation)

$$\delta_{NO_3} = \varepsilon \ln f + \delta_{NO_3 - \text{initial}}$$
  
(f = C/C<sub>0</sub>)



# Column experiments

# Isotope effects during denitrification

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$$\delta_{NO_3} = \varepsilon \ln f + \delta_{NO_3 - \text{initial}}$$





5 identical batches with aquifer sediment from two different drilling cores 140 g sediment + 177 mL synthetic groundwater (NO<sub>3</sub>: 100 mg/l, SO<sub>4</sub>: 125 mg/L)



#### nitrate concentrations





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#### sulfate concentrations





5 identical batches with aquifer sediment from two different drilling cores 140 g sediment + 177 mL synthetic groundwater (NO<sub>3</sub>: 100 mg/l, SO<sub>4</sub>: 125 mg/L)



### sulfate concentrations and isotopes



# Insights from isotopic composition of sulfate



Autotrophic denitrification seems to be the prevailing pathway during experiments despite TOC >> FeS<sub>2</sub>



# Insights from isotopic composition of sulfate



Autotrophic denitrification seems to be of major importance in the aquifer despite TOC >> FeS<sub>2</sub>



# **Implication for Water Framework Directive**

# Meeting the targets of the WFD (good chemical status of groundwater) at the groundwater surface:

Time frame is dependent on (i) time needed to change agricultural practices <u>plus</u> (ii) travel time of leachate through the unsaturated zone.

In the Hessian Ried targets could be met at the groundwater surface e.g. for 17% of the area within 3 years after changing agricultural practices and for 10 % in more than 15 years.

For some monitoring wells of the WFD with deeper screens, concentrations will be high for decades to come or will even increas due to exhausted reduction potential on a flowpath.

> Travel times for leachate through the unsaturated zone Modell: MIKE-SHE (BGS Umwelt)



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## Part II: The Selke Story

N. Trauth et al. (UFZ Department Hydrogeology), K. Knöller

mass twin





# Impact of groundwater/ surface water exchange on the fate of nitrate in groundwater





# Installations

0 100 200 300 400 500 m

#### Surface water monitoring

- Groundwater wells in riparian zone
- Groundwater wells in regional groundwater

G

H

Sauerbach tributary F

# **Hydrochemistry - Nitrate**



#### Inventory: Background isotope signatures



No obvious difference between river nitrate and groundwater background

any change in isotope signatures related to biogeochemistry



#### Temporal variation of **nitrogen** isotope signatures in nitrate



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Validation of denitrification in biogeochemical hot spots:



Expected parameters:

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Validation of denitrification in biogeochemical hot spots:

![](_page_33_Figure_2.jpeg)

Expected parameters:

![](_page_34_Figure_0.jpeg)

# Tracing the fate of nutrients in agricultural catchments by stable isotope techniques

![](_page_35_Figure_2.jpeg)

## Part III: The Bode Story

Christin Müller, Matthias Zink, Luis Samaniego, Ronald Krieg, Michael Rode, Ralf Merz, Kay Knöller

![](_page_35_Picture_5.jpeg)

![](_page_35_Picture_6.jpeg)

![](_page_35_Picture_7.jpeg)

![](_page_35_Picture_8.jpeg)

#### Isotope monitoring in the Bode river catchment

![](_page_36_Figure_1.jpeg)

#### Nitrate isotopes along the main river

#### **Spatial distribution**

![](_page_37_Figure_2.jpeg)

Mueller et al. (2016, ES&T)

Nitrate isotope fingerprinting in the mesoscale Bode River catchment: temporal variability of nitrate concentrations

![](_page_38_Figure_1.jpeg)

Mueller et al. (2016, ES&T)

#### Nitrate isotope fingerprinting in the mesoscale Bode River catchment:

Regional relevance and variability of N-sources and sinks

![](_page_39_Figure_2.jpeg)

#### Nitrate isotope fingerprinting in the mesoscale Bode River catchment:

Regional vs. local impact of denitrification

#### Potential denitrification vectors

![](_page_40_Figure_3.jpeg)

Thank you for your attention!

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# Denitrification pathways in the aquifer

![](_page_42_Figure_1.jpeg)

# Denitrification pathways in the aquifer

![](_page_43_Figure_1.jpeg)