

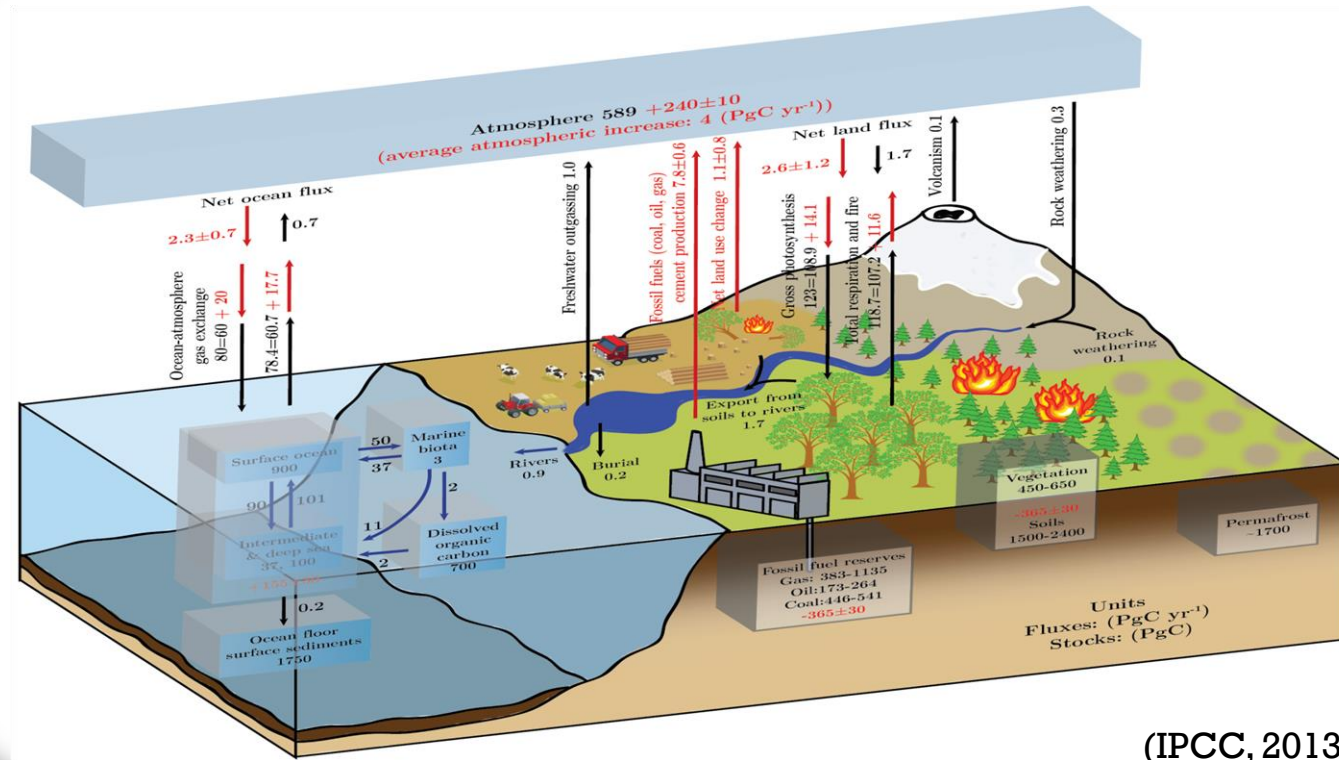
A photograph of a cave interior, showing a variety of stalactites and stalagmites. The cave floor is covered with numerous stalagmites of different heights and shapes. The walls and ceiling are also covered with stalactites, some of which are quite long and thin. The lighting is dim, with a few bright spots where the cave floor is illuminated, creating a dramatic effect. The overall atmosphere is dark and mysterious.

Assessing sources of cave air CO₂

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mass
TWIN

Carbon cycle

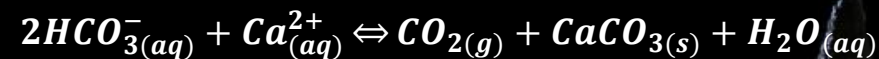
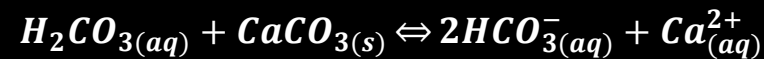
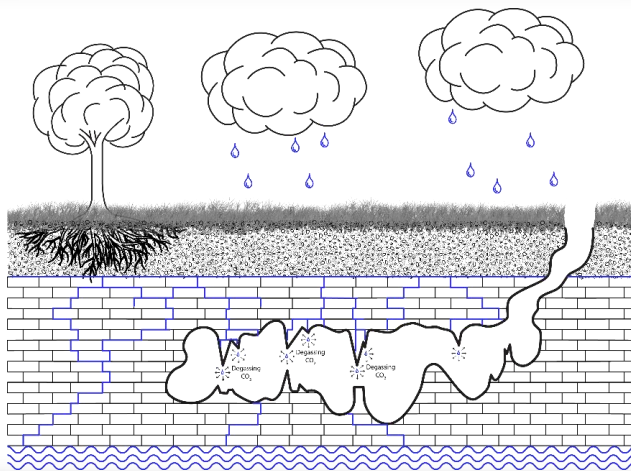
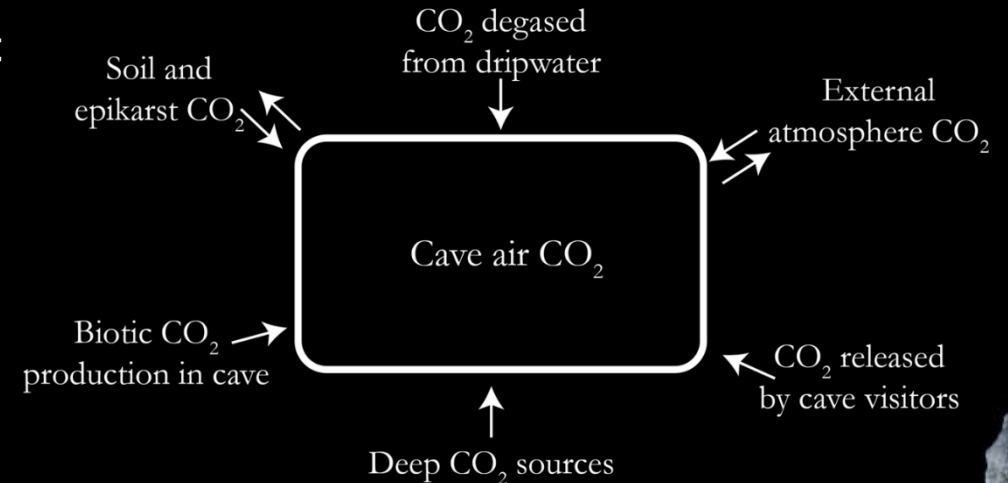


The carbon cycle is the set of biogeochemical processes by which carbon undergoes chemical reactions, changes form, and moves through different reservoirs on earth, including living organisms.

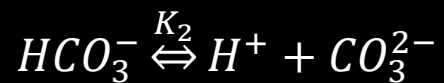
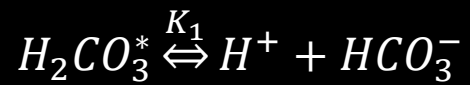
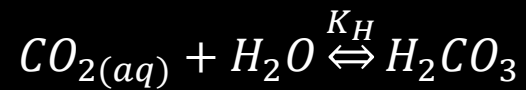
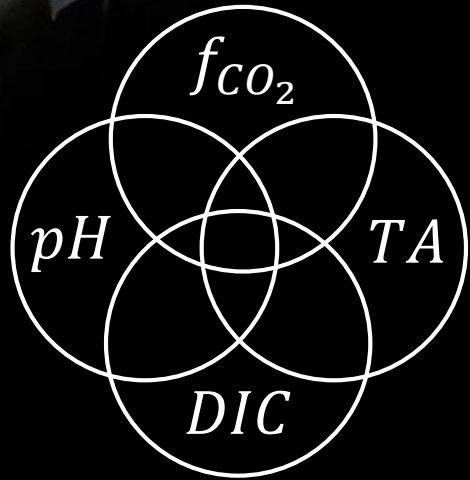
CO₂ in caves

CO₂ accumulating in caves can come from different sources through different pathways:

- External atmosphere
- Directly from overlaying soil and epikarst
- Degassing from dripwaters
- Cave visitors,
- in-cave biotic production,
- deep sources

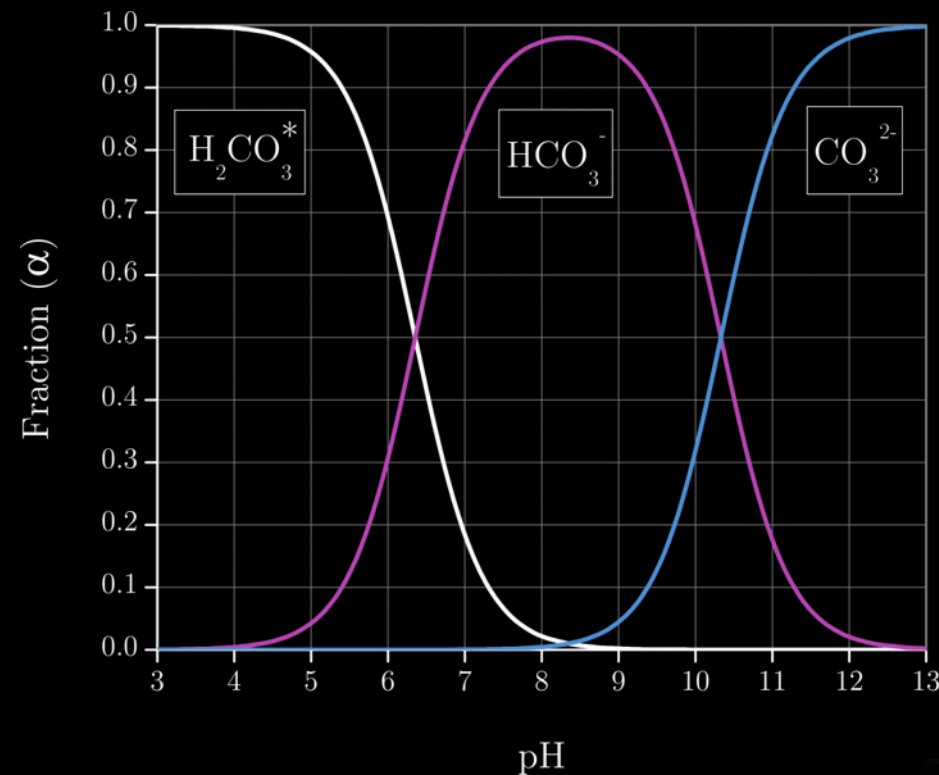


Carbonate equilibrium



$$DIC = [H_2CO_3^*] + [HCO_3^-] + [CO_3^{2-}]$$

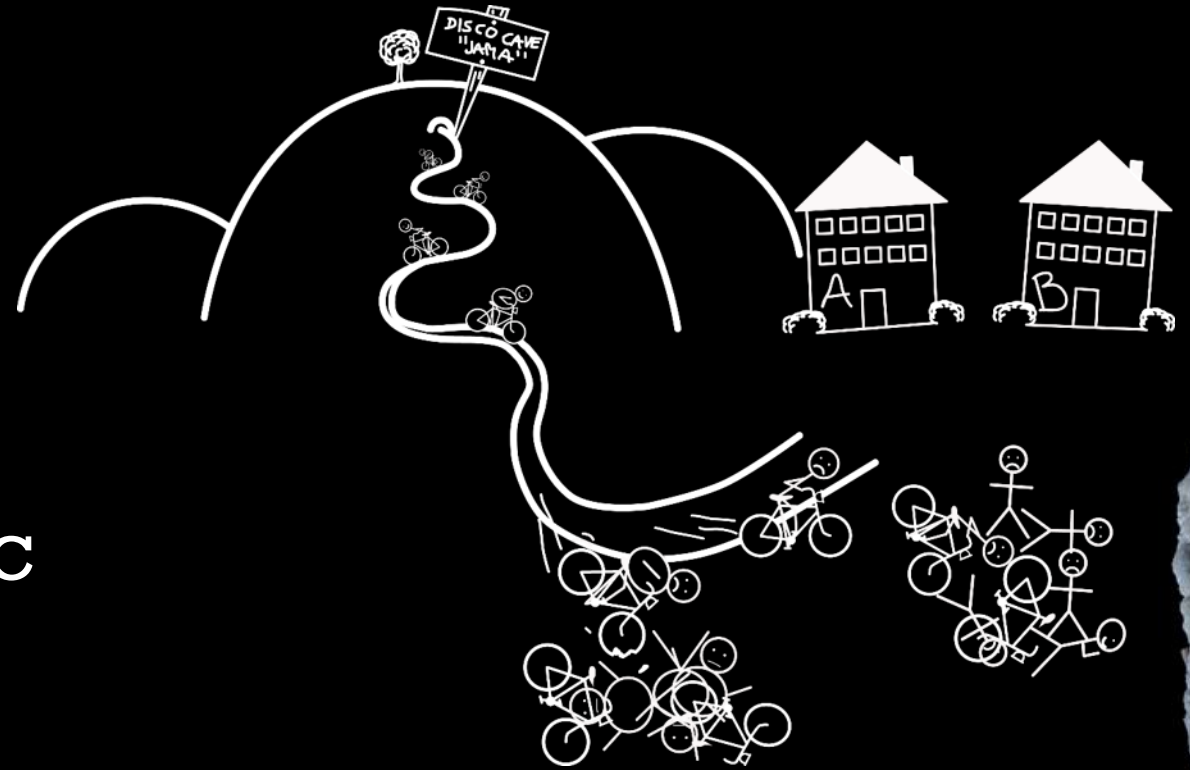
$$TA = [HCO_3^-] + 2[CO_3^{2-}] + [OH^-] - [H^+] + \dots$$



Source assessment

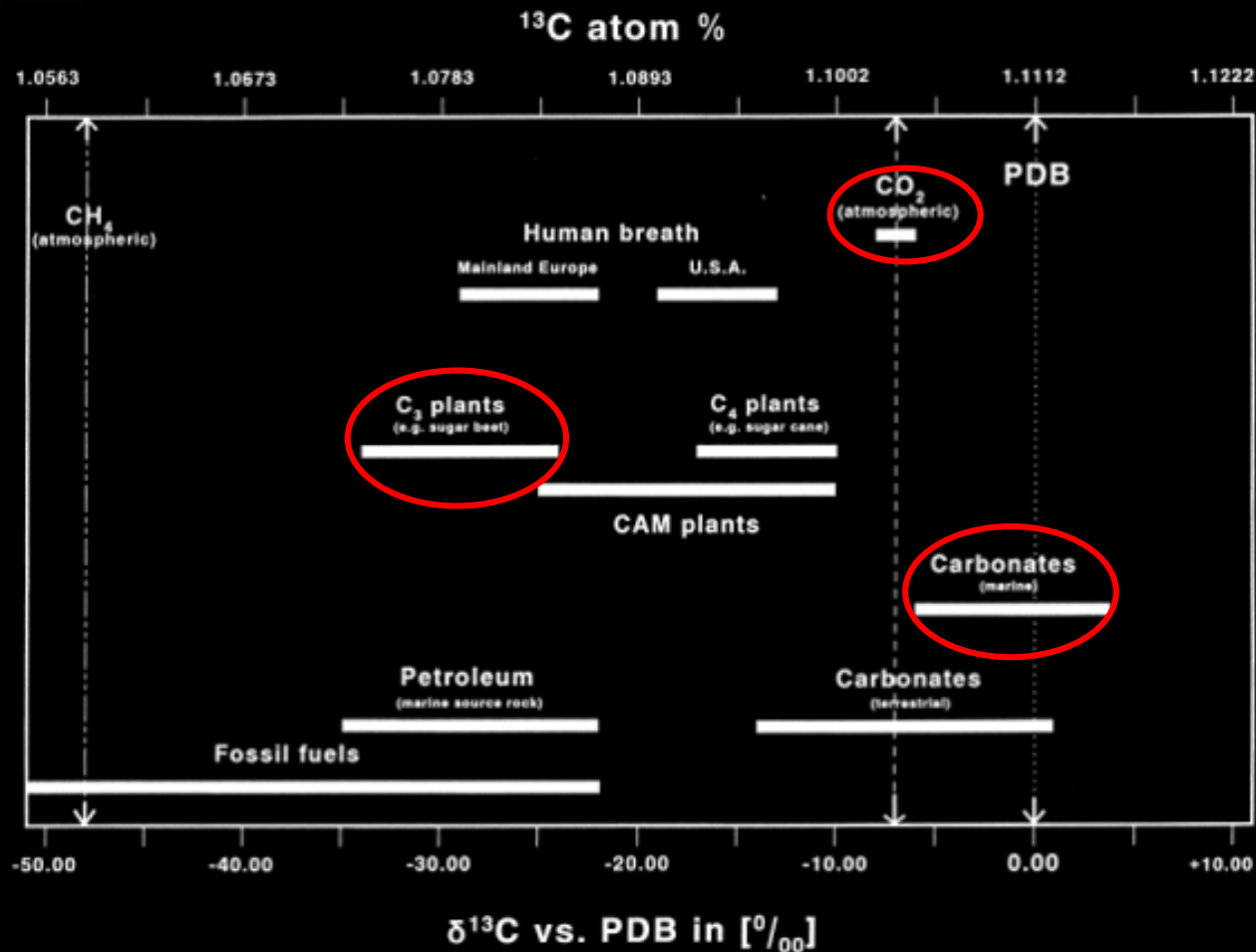
Use of tracers:

- Stable carbon isotopes, $\delta^{13}\text{C}$
- Radiocarbon, $\Delta^{14}\text{C}$
- Radon, ^{222}Rn



Stable carbon isotopes ($^{13}\text{C}/^{12}\text{C}$)

$\delta^{13}\text{C}$ - a useful tool to study the carbon cycle

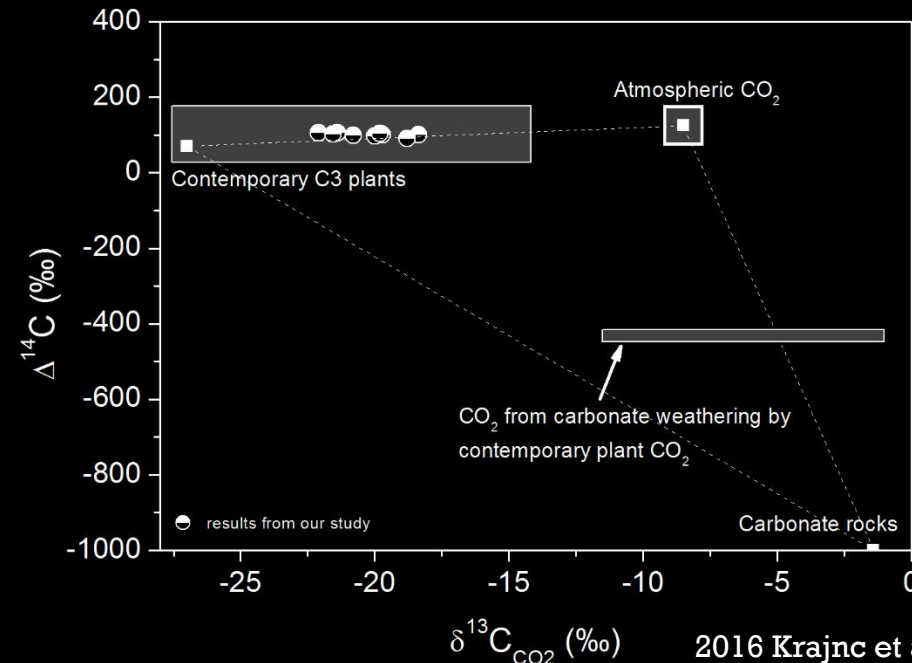


Some typical examples of natural $\delta^{13}\text{C}$ values grouped according to origin along the scale of ^{13}C natural abundance (Meier-Augenstein, 1999)

Radiocarbon (^{14}C)

- Unstable with half-life of 5730 ± 40 years
- Enters plant and animal life through photosynthesis and food chain
- After the death of the organism only ^{14}C decay (basis for ^{14}C dating)
- Useful to distinguish between „young“ and „old“ sources (e.g. atmospheric CO_2 and C from carbonate bedrock)

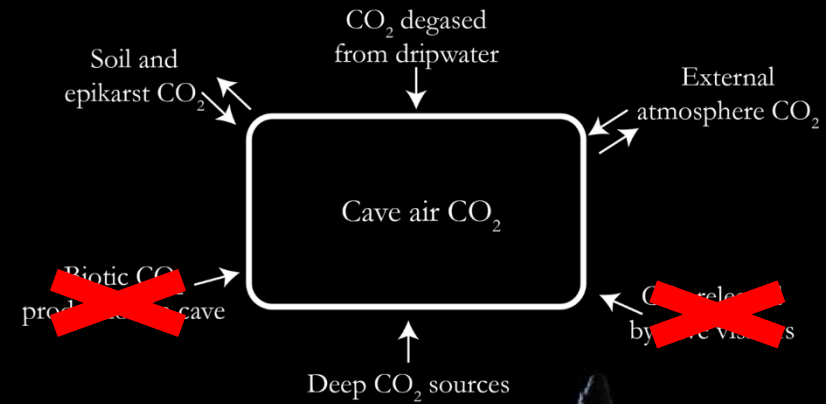
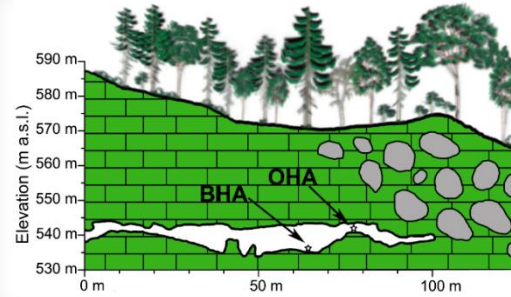
We used it as the second tracer which enabled us to partition between up to 3 carbon sources:



Radon (^{222}Rn)

- Nobel gas (chemically inert)
- Radioactive with half-life of 3.82 days (easy to detect)
- Caves can have high radon concentrations (low ventilation with constant emissions from walls, water etc.)
- If CO_2 is the carrier gas a significant correlation exists between both gasses
- Investigation of cave ventilation and movement of cave air CO_2 (carrier gas)

Sources of cave CO₂



Methods

Sampling and measurements:

AIR (soil and cave):

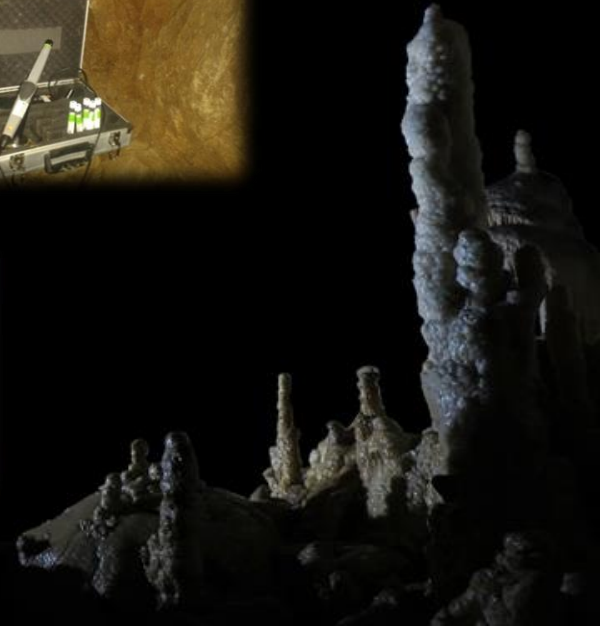
- CO₂ concentration: NDIR probe (TESTO-453)
- Carbon stable isotopes: IRMS (Europa scientific 20-20)
- $\Delta^{14}\text{C}$ (sampled with NaOH solution, measured at AMS in JAEA, Mutsu, Japan)
- ²²²Rn: scintillation cells + alpha counter (PRM-145; AMES)

SOC and parent rock material:

- IRMS (Europa scientific 20-20)

METEOROLOGICAL DATA:

- Meteorological data form *in-situ* and/or
- Nearby meteorological station (ARSO)



Methods

Sampling and measurements:

WATER (soil and dripwater):

- DIC (calculated from T, pH and Alkalinity)
- $\delta^{13}\text{C}_{\text{DIC}}$ (IsoPrime IRMS + MultiFlow Bio module)
- Cation concentrations (ICP-OES)
- Anion concentration (ion chromatography)
- pH (Corning 315 pH Meter + Orion Ross pH electrode)
- Conductivity (Myron Ultrameter IITM)
- Alkalinity (Gran titration)



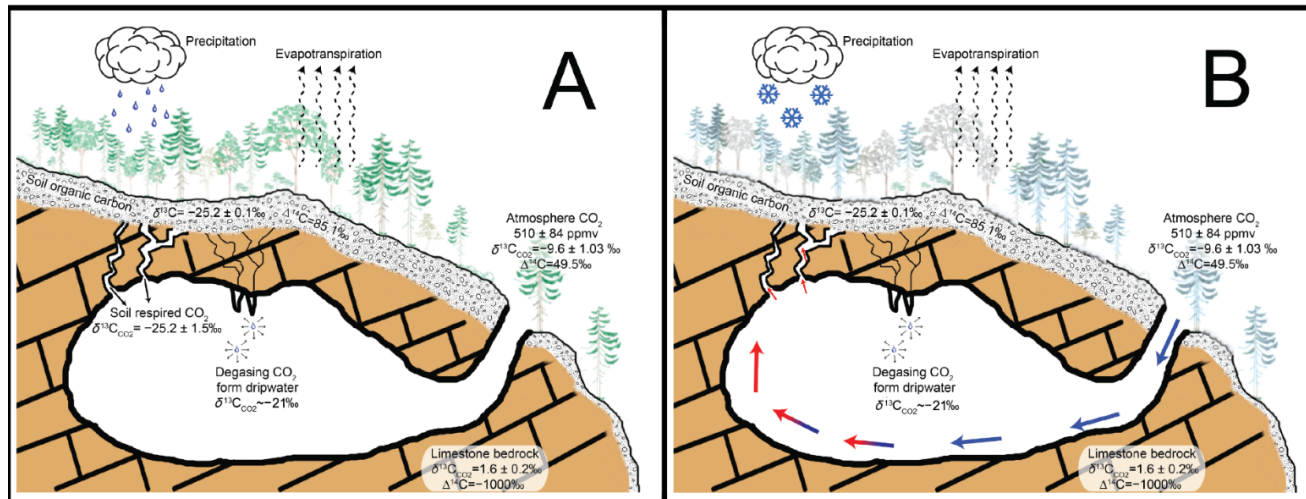
DATA MANIPULATION AND MODELING:

- Raw data manipulation and statistical analysis (R statistical environment)
- Modeling of the carbonate equilibrium and isotopic fractionation (PHREEQC and MIX4)

Sources of cave CO₂

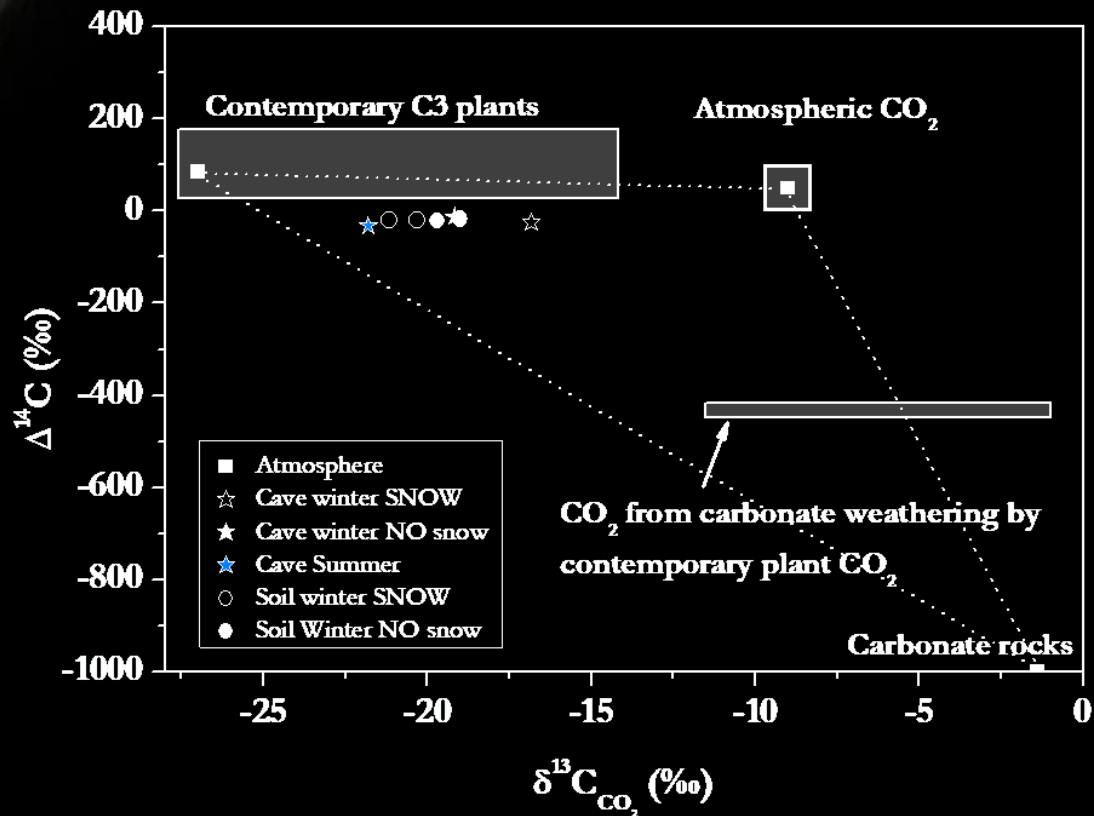
Results:

- Outside temperature is the main factor affecting cave ventilation
- During snow cover and freezing temperatures, the ventilation is inhibited
- Good correlation between CO₂ and ²²²Rn (common driving force)

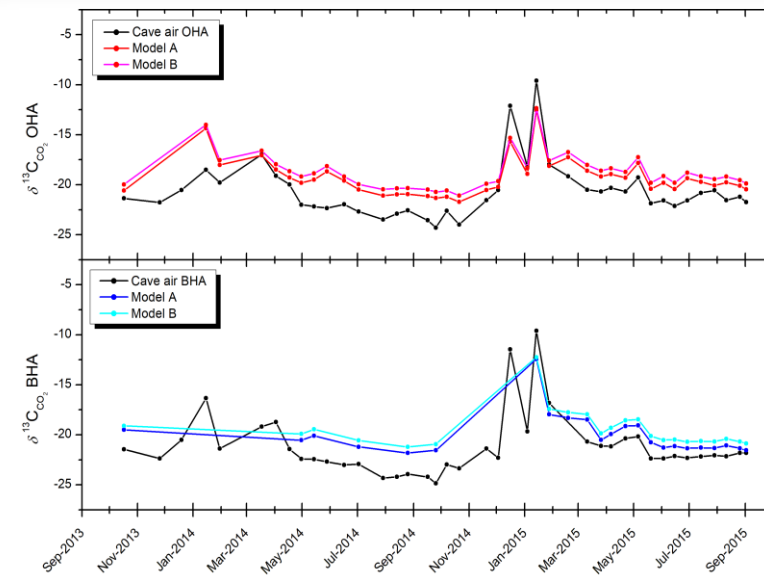


Results: Sources of cave CO₂

- Majority of cave CO₂ indeed comes from soil respiration
- Comparison of modelled values with measured values suggests that CO₂ degassing from dripwaters can not be a major CO₂ source

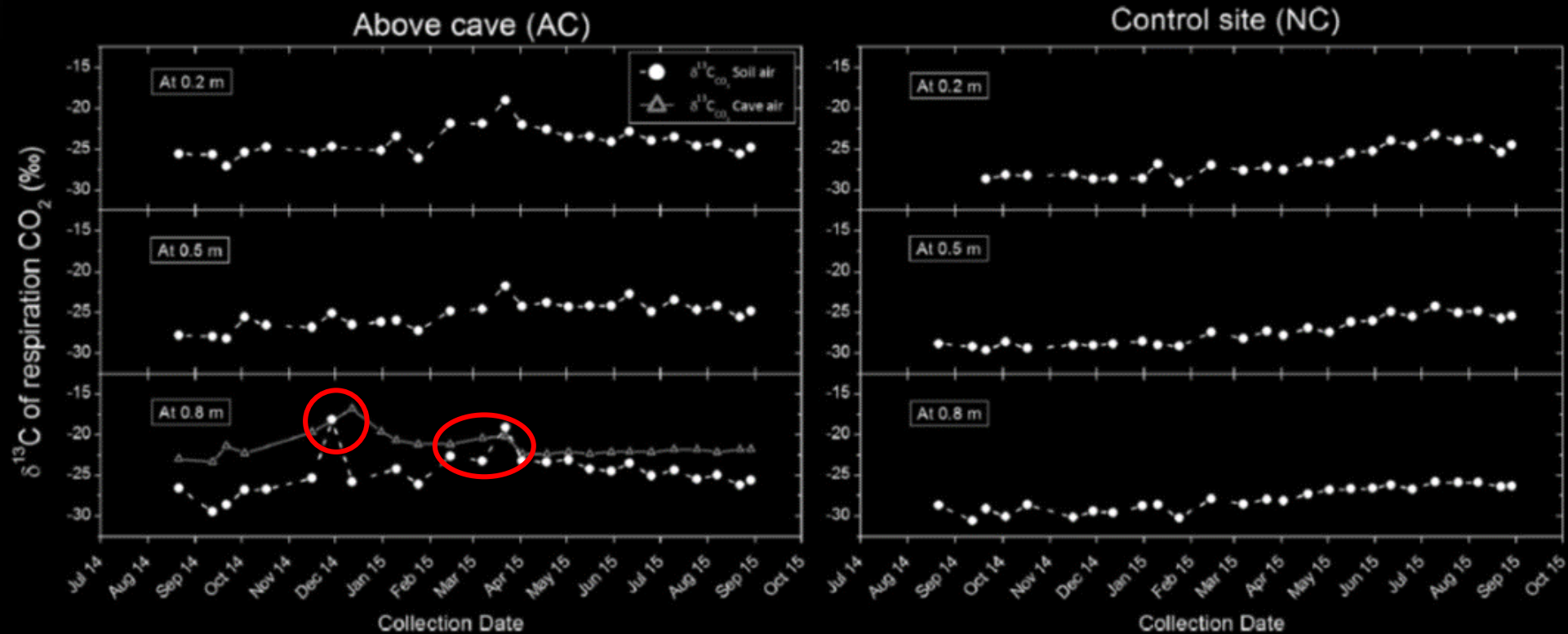


Krajnc et al., (in press) Geochim. Cosmochim Acta



Soil CO₂ sources above the subterranean cave

There was an additional source of CO₂ present in the soil above the cave during the cold periods which coincided with the period of cave ventilation.



(2016, Krajnc et al., J Soils Sediments)

Conclusions

The levels of CO₂ concentrations, $\delta^{13}\text{C}_{\text{CO}_2}$ and ^{222}Rn in the cave air have typical seasonal pattern

These values are not primarily controlled by soil CO₂ productivity, but are mainly caused by the ventilation

Major source of cave CO₂ in the studied chamber is the CO₂ originating from soil respiration, transported directly to the cave,

The findings of this study demonstrate that isotope geochemistry is a very useful tool to address the sources of CO₂.

Acknowledgements

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A photograph of a cave interior, featuring numerous stalactites hanging from the ceiling and stalagmites growing from the floor. The cave is dimly lit, with light reflecting off the wet, mineral-rich surfaces of the rock formations. The formations vary in size and shape, creating a complex and textured environment.

Thank You

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