



International Atomic Energy Agency

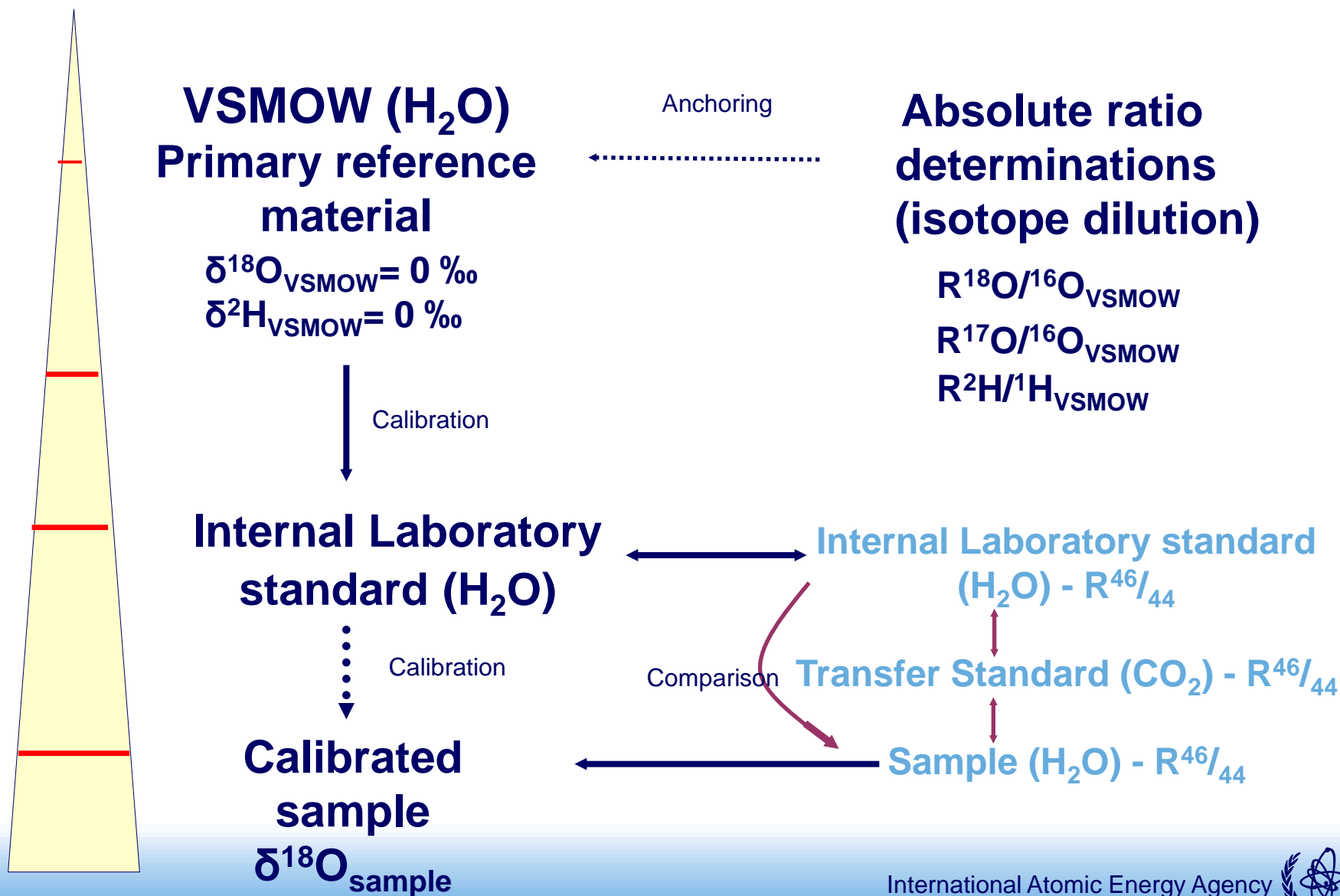
Basic concepts and the Use of Reference Materials in Stable Isotope Metrology

Manfred Gröning

Terrestrial Environment Laboratory, IAEA

MASSTWIN Training course on Metrology, Ljubljana, 7-9 Dec 2016

Traceability & Uncertainty pyramid



The problem in short

The mass spectrometer is installed and tested

How to ensure it is fit for the work ahead ...

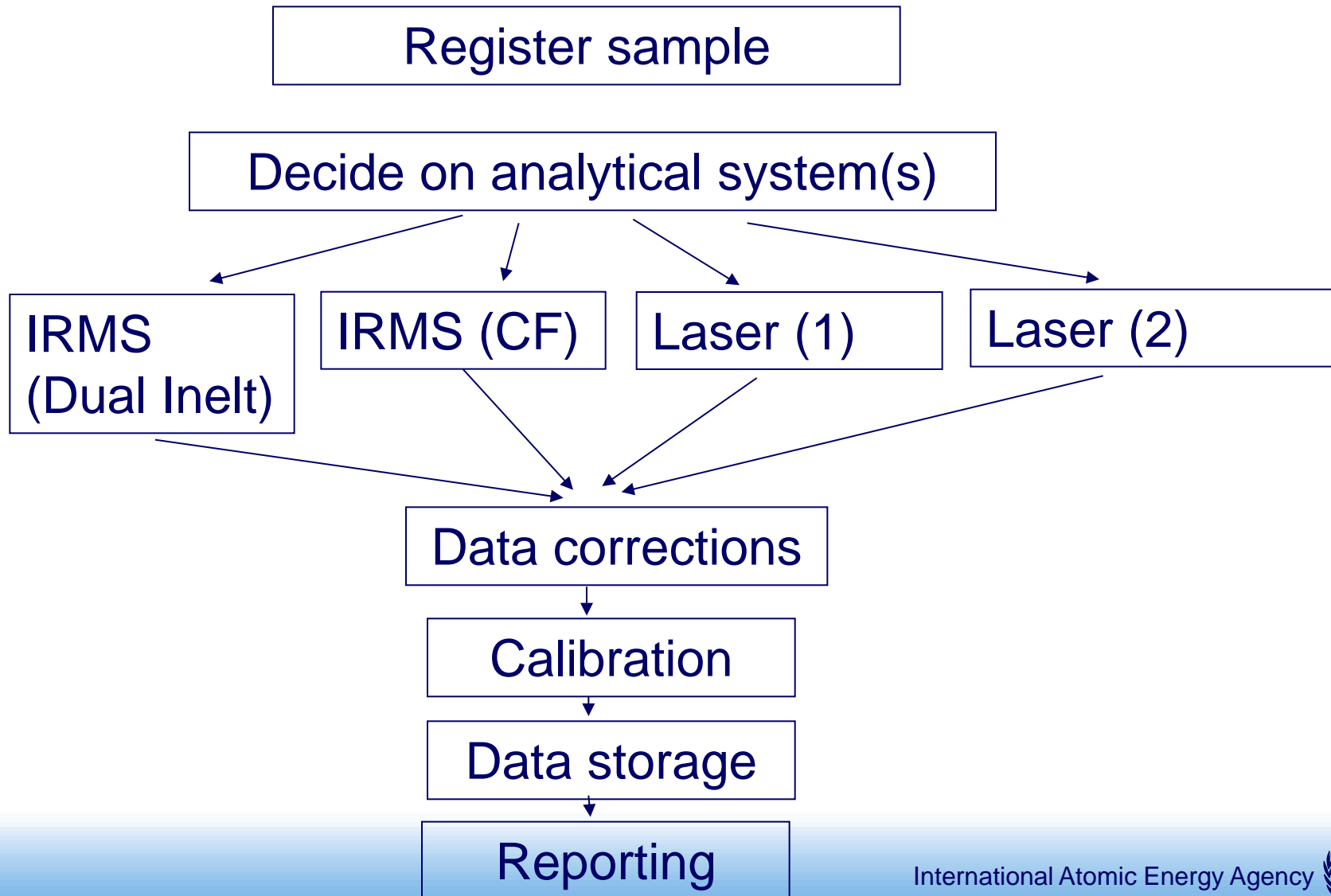


Basic Concepts

After installation and acceptance testing of a new MS:

- 1. Decide on internal laboratory standards (ILS), obtain and measure them to ensure long term performance**
- 2. Calibrate the ILS samples versus suitable reference materials (RMs) and monitor data (QC, ILS)**
- 3. Establish routine reporting**

Sample flowchart at a Laboratory



1. Internal Laboratory Standards (ILS)

- International measurement standards VSMOW2, SLAP2
- Internal laboratory standards (ILS)
- Unknown samples

Why intermediate step ILS necessary?

- VSMOW2 and SLAP2 - small quantities.
Do not use them for everyday calibrations!
- Need for an inexpensive set of internal standards that are calibrated versus VSMOW2 and SLAP2.
- The internal standards are the entire basis for the accuracy of your results!!!

⇒ They are as important as the instrument!

Internal Laboratory Standards (ILS)

- Use in daily calibrations of routine measurements
- Several ILS standards required (at least 3), to be stored in suitable condition in sufficient amounts for at least a decade of lab operation
- Cover full isotopic range of routine samples, should have similar chemical properties
- Calibration of ILS is of utmost importance as it defines possible accuracy of any measurement; regular re-calibrations recommended

Internal (in house) laboratory standards

- **Where do I get internal standards?**
- **Not many suppliers, only sold in small quantities, and they are expensive.**
- **The best way is to make your own!**
- **This is what all the good isotope laboratories do, and it is not difficult.**
- **Following example for water – generally applicable for all other applications ...**

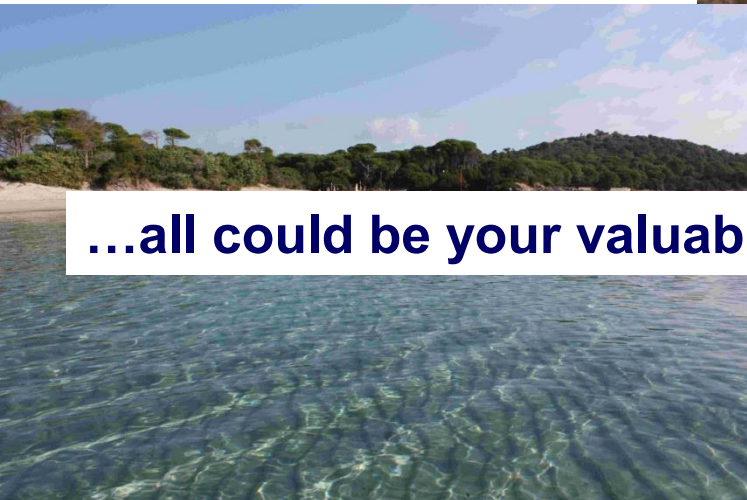
Possibilities



What do these pictures of water have in common ?



...all could be your valuable future internal lab standards ...



Making internal lab standards

- You want at least 3 ILS standards
- Two as calibration standards
(a “heavy” standard and a “light” standard bracketing the range of typical sample values)
- One quality control standard (QC)
(QC isotopically intermediate)
- Try to conserve them (filtrate, distill/sterilize ...)
- Additional standards useful (e.g. a very heavy standard for highly evaporated samples). A “library” of standards is nice, but not always essential.

Internal laboratory standards - II

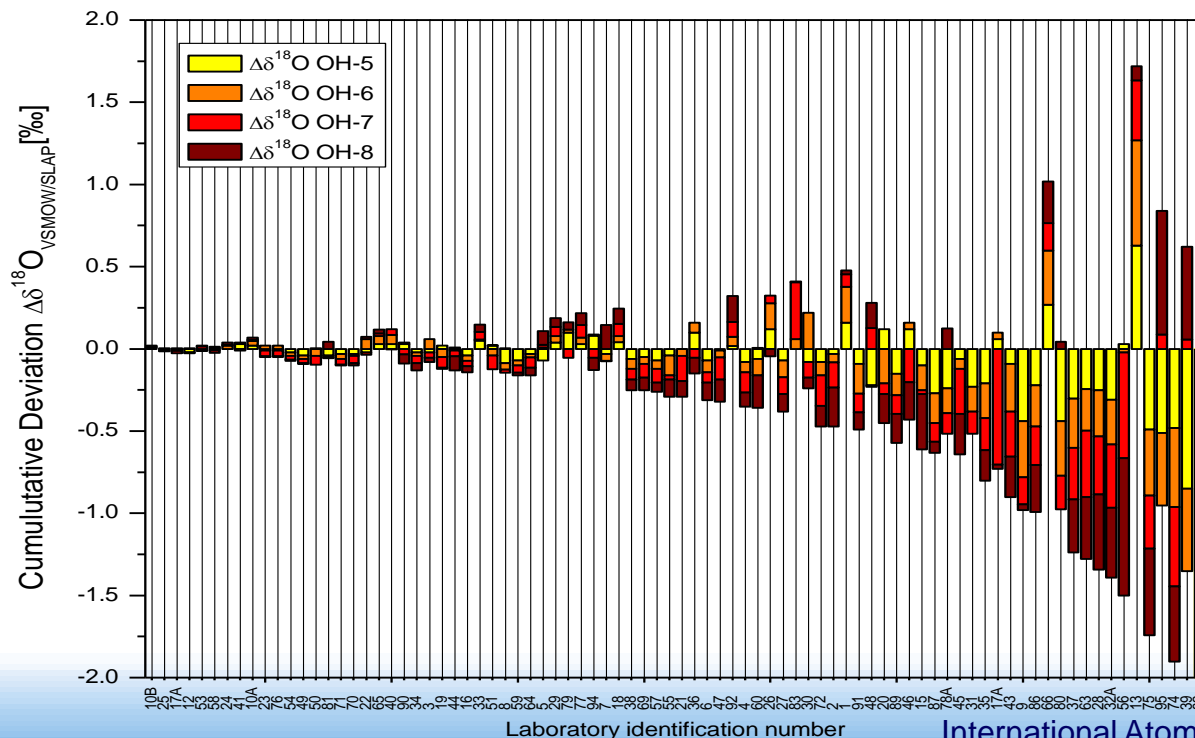
Main problem: Calibration of standards

- Ensure accuracy of produced data, quantify precision
- careful and proper calibration of the isotopic composition
- Result:
 $\delta^{18}\text{O}(\text{std11}) = 0.07 \pm 0.03 \text{ ‰ vs. VSMOW/SLAP}$
- Question: how to quantify uncertainty?

Internal laboratory standards - III

Second major problem: Storage of standards

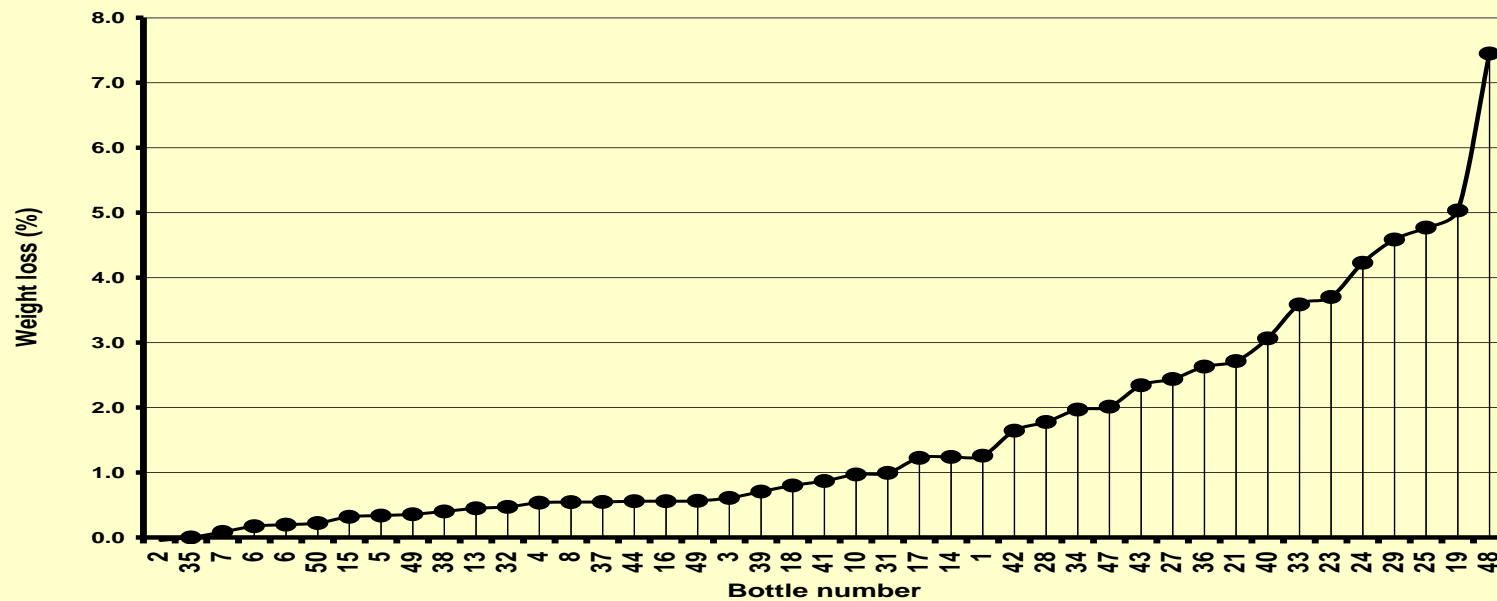
- Avoid any evaporation or fractionation during storage
- Remember: you will not be able to recognize it easily!



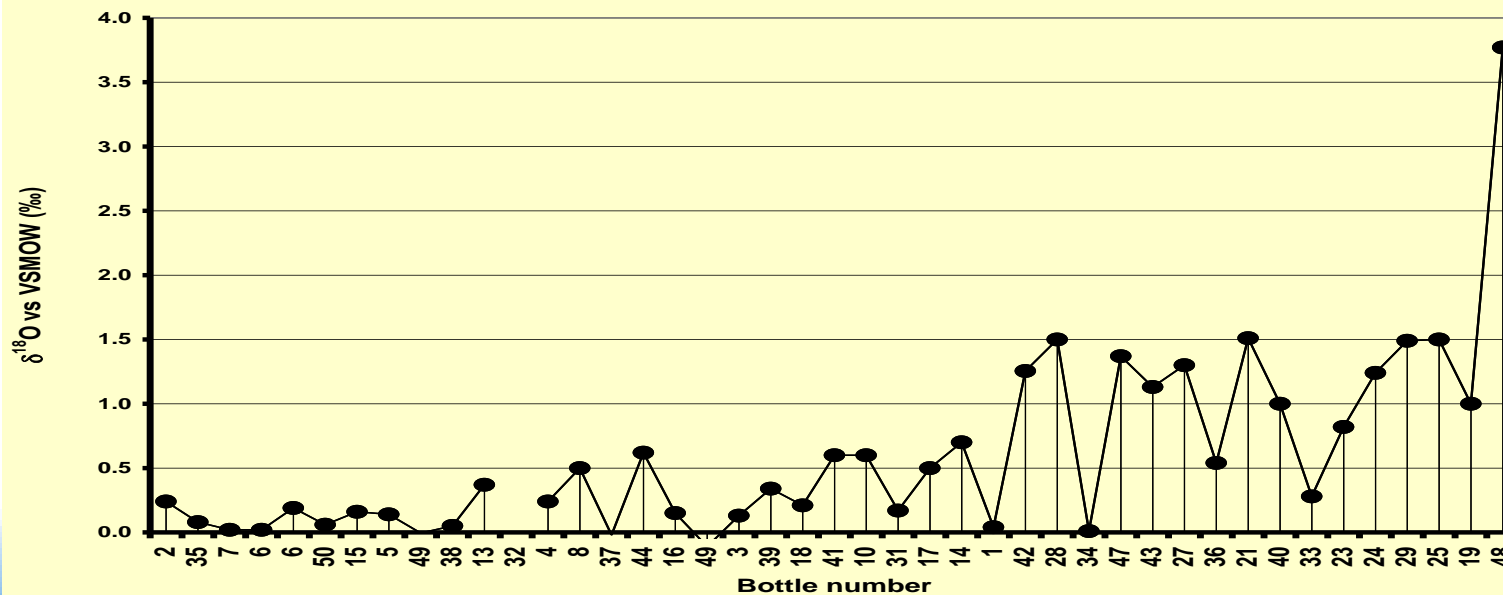
WICO2002



Changes in weight after 12 months storage



$\delta^{18}\text{O}$ values after 12 months storage



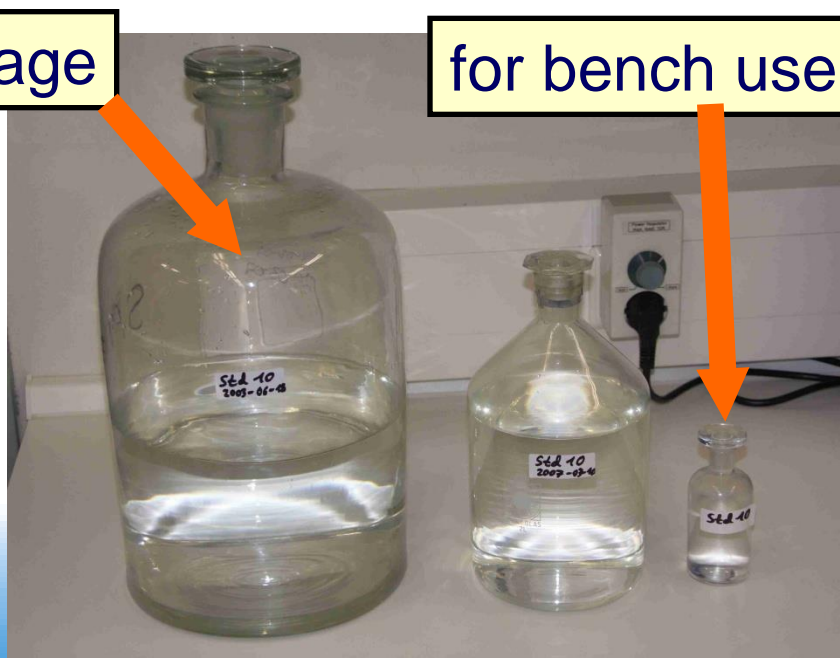
Internal laboratory standards IV

Storage containers:

- Metal barrels under gas overpressure preferred choice
- Glass flasks and bottles: large/medium/small
- (See IAEA-IHL TPN43 report)



for storage



for bench use



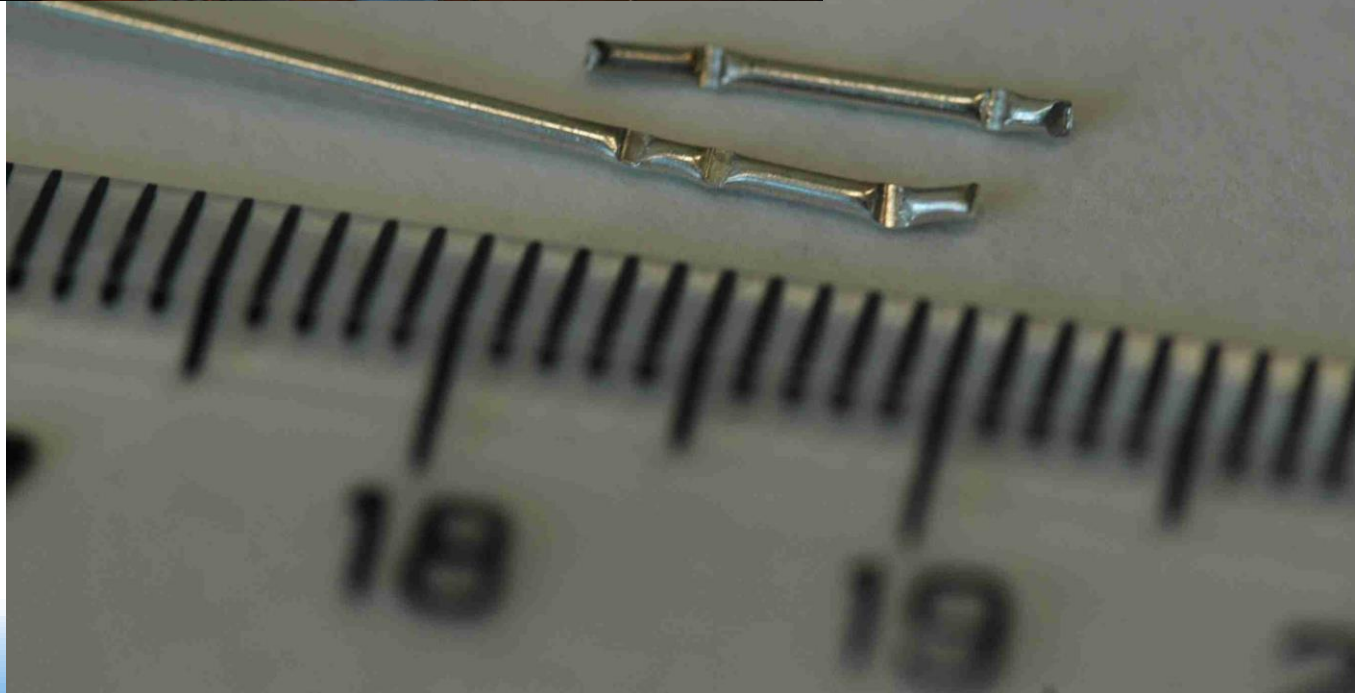
Water distillation

Photo of a standard laboratory distillation system being used for preparation of an internal standard.



Preparation of water standards using the “solid water” method

- For analysis of materials different from water for H and O isotopes, calibration of solid ILS may be complicated. Without suitable RMs calibration has to be done versus VSMOW2 water.
- The major problem is the evaporation of water during sample preparation before analysis.
- A new method was developed to avoid this problem.
- Suitable amounts of water is sealed in silver capillaries, which can be used together with solid samples in an autosampler.
- **See Qi et al. 2009**



Considerations for solid ILS materials

- For routine analysis of other materials (organics, inorganics like carbonates), internal laboratory standards are needed as well.
- Similar considerations apply:
- Carbonates: care for isotopic exchange, dry storage
- Organic materials: long term stability, side effects (caffeine - N), isotopic compositions
- Calibration of these solid standards has to be performed against suitable reference materials
- Identical Treatment principle – if possible

2. The Calibration process

- Perform calibration of ILS versus VSMOW2 and SLAP2 only under the best instrumental conditions
- VSMOW2 and SLAP2 are precious, don't waste them!
- Perform enough measurements to derive statistically well founded results
- Repeat on several days
- Calibration Excel template available (**CalibTemplate.xls**)
- Have other good isotope laboratories analyze your internal standards as an independent check

Calibration of measurement data

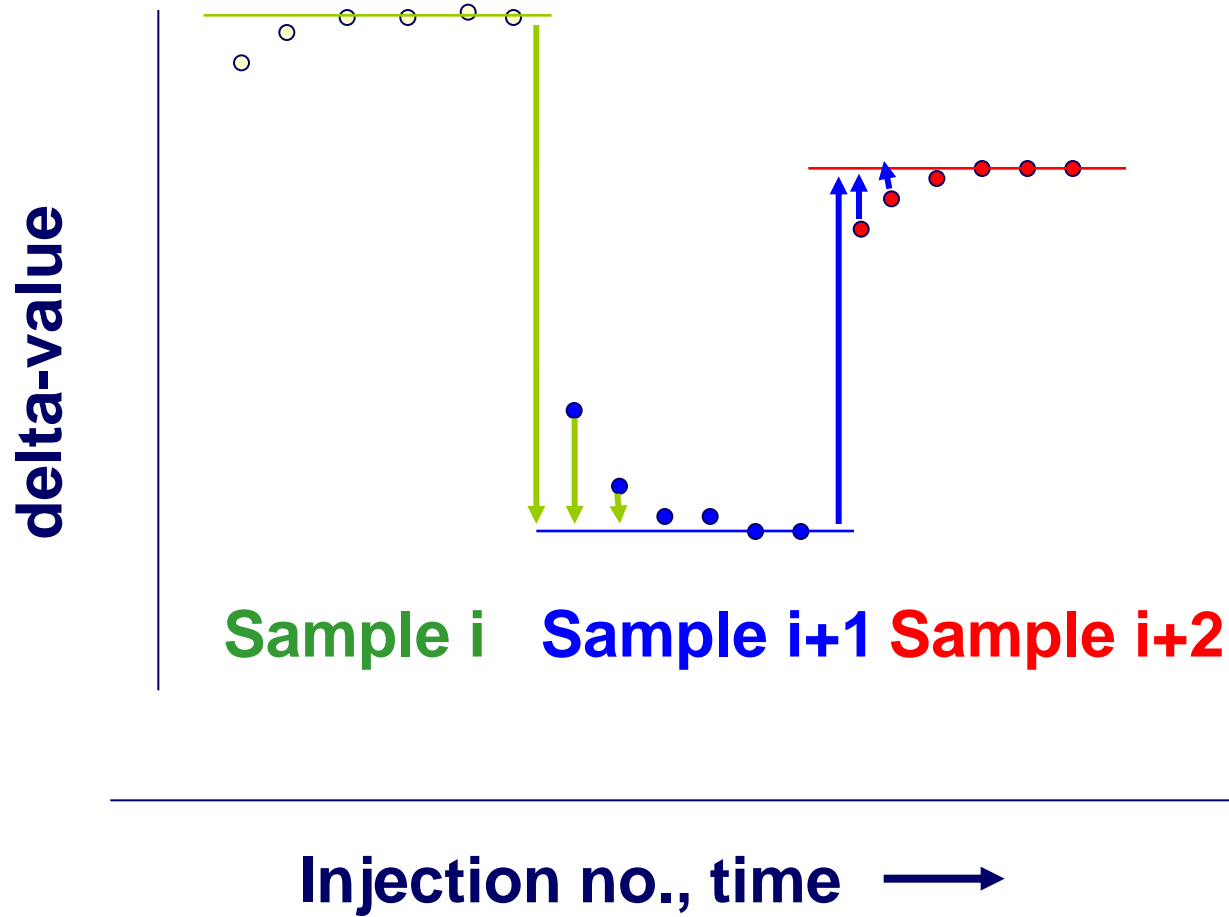
Calibration of measurements is key to ensure comparability of data between laboratories

- a) Raw data as measured by instrument**
- b) External influences and corrections – temperature fluctuations, variations in used amounts, background**
- c) Memory effect correction**
- d) Drift of isotope data with time and correction**
- e) Actual calibration using ILS and**
- f) Uncertainty evaluation**

2.a Raw data

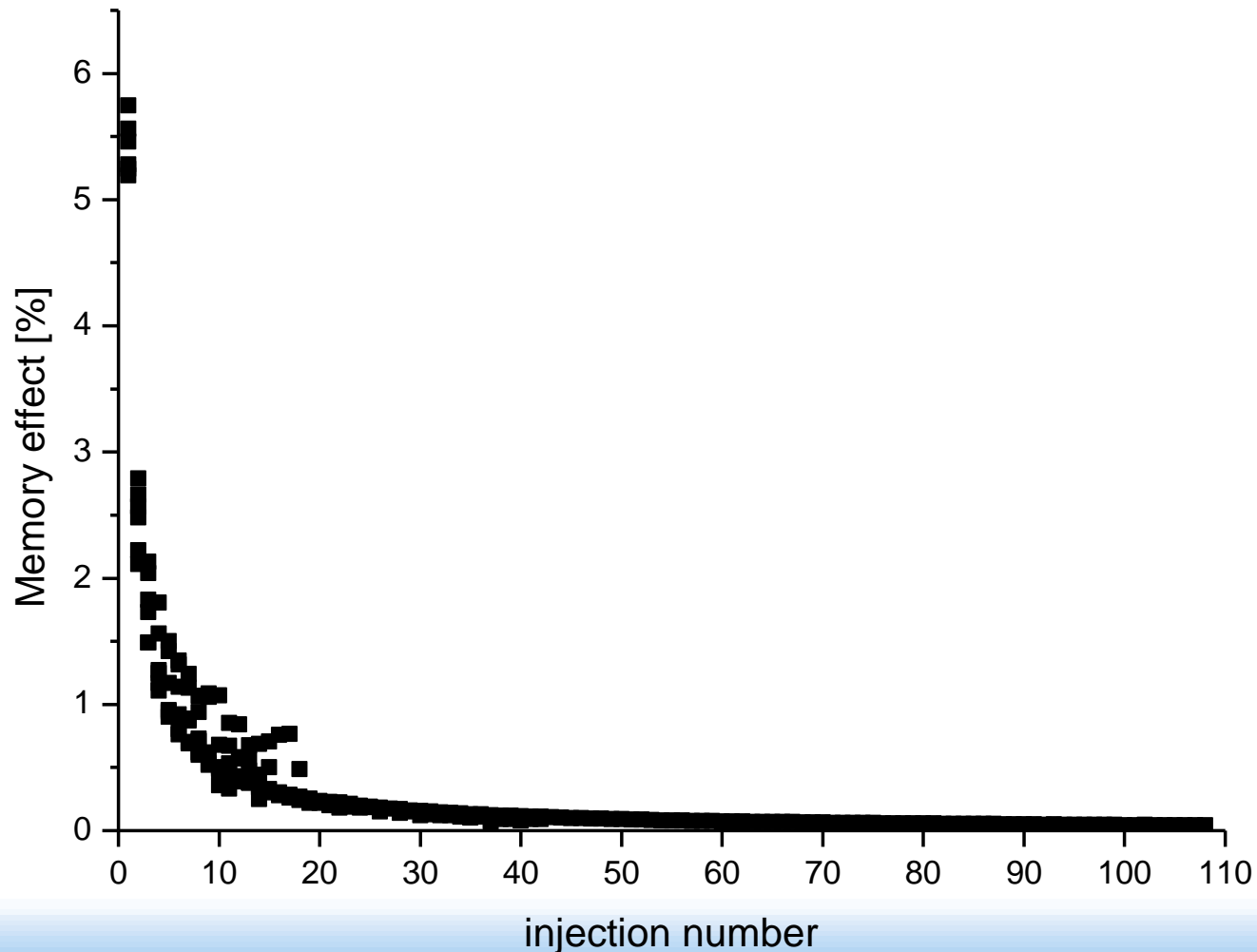
- Raw data as provided by instrument (see **Gonfiantini 1981 IAEA TRS No.210; Allison 1995, IAEA TECDOC-825**)
- point of view of a modern user of today
- Correction and calibration of data is handled differently by each instrument provider, consistency can be improved by using a common calibration mechanism:
- SICalib – Provided free of charge by IAEA
https://nucleus.iaea.org/rpst/referenceproducts/Analytical_Methods/Stable_Isotope_Reference_Laboratory/index.htm
- LIMS provided free of charge by T. Coplen at USGS
<http://water.usgs.gov/software/LIMS/>

2.c Memory correction

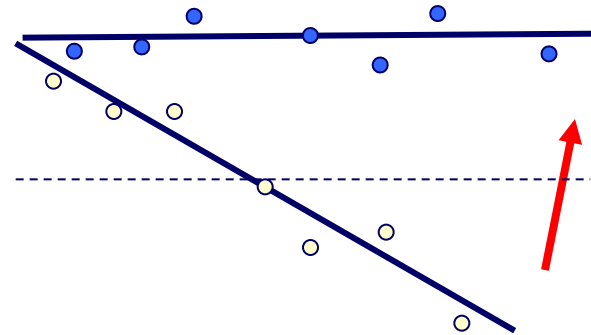


Instrument Memory for Laser (in % of isotopic difference)

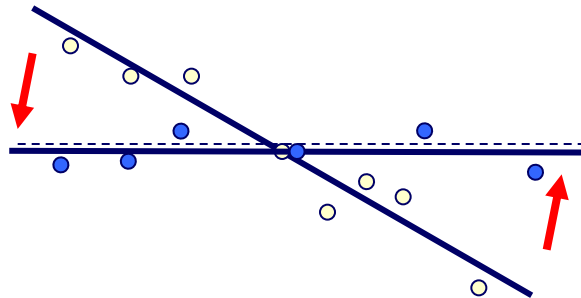
Memory for $\delta^2\text{H}$ measurements



2.d Drift correction



Drift corrected
mean
Mean **Incorrect !!!**



Mean & Weighted
Correct

For further details of the drift correction and used formula see **Manual of SICalib**

2.e Calibration

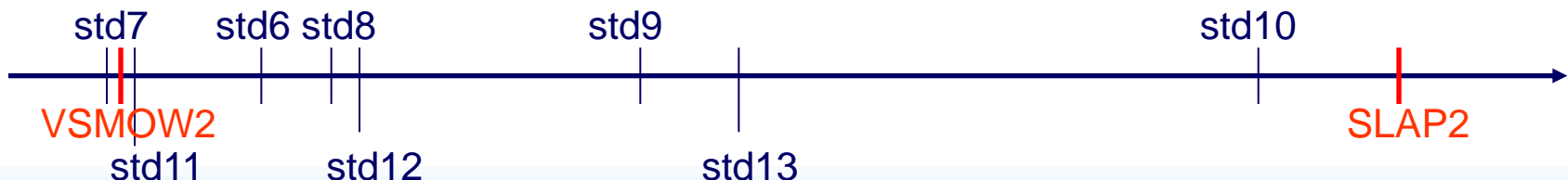
Basic equation for 2-point calibration:

$$\delta^2\text{H}_{\text{sample}} = \delta^2\text{H}_{\text{cal1}} + \left(\delta_{\text{W}}^2\text{H}_{\text{sample}} - \delta_{\text{W}}^2\text{H}_{\text{cal1}} \right) \cdot \left(\delta^2\text{H}_{\text{cal2}} - \delta^2\text{H}_{\text{cal1}} \right) / \left(\delta_{\text{W}}^2\text{H}_{\text{cal2}} - \delta_{\text{W}}^2\text{H}_{\text{cal1}} \right)$$

with cal1=VSMOW2, cal2=SLAP2, data indicated with W versus working standard, all other data calibrated on the VSMOW/SLAP scale.

In first instance Internal Laboratory Standards have to be calibrated.

At IAEA a series of water standards is available:



plus a series of 27 secondary internal laboratory standards ISL-01 to ISL-27

(Calib Template.xls)

$\delta^{18}\text{O}$ results vs VSMOW-SLAP in ‰ (calibrated & normalised on VSMOW-SLAP scale) - listed in

	VSMOW2	SLAP2	xx2	Std6	Std8	GISP/Std9	Std10	Std11
Average	0.000	-55.500	#DIV/0!	-8.621	-11.291	-24.773	-50.824	0.078
std.dev.	0.043	0.054	#DIV/0!	0.033	0.031	0.027	0.044	0.049
no.	6	5	0	4	5	5	4	6

Correction factor for VSMOW-SLAP scale =	0.979313	long-term lab reproducibility [‰]
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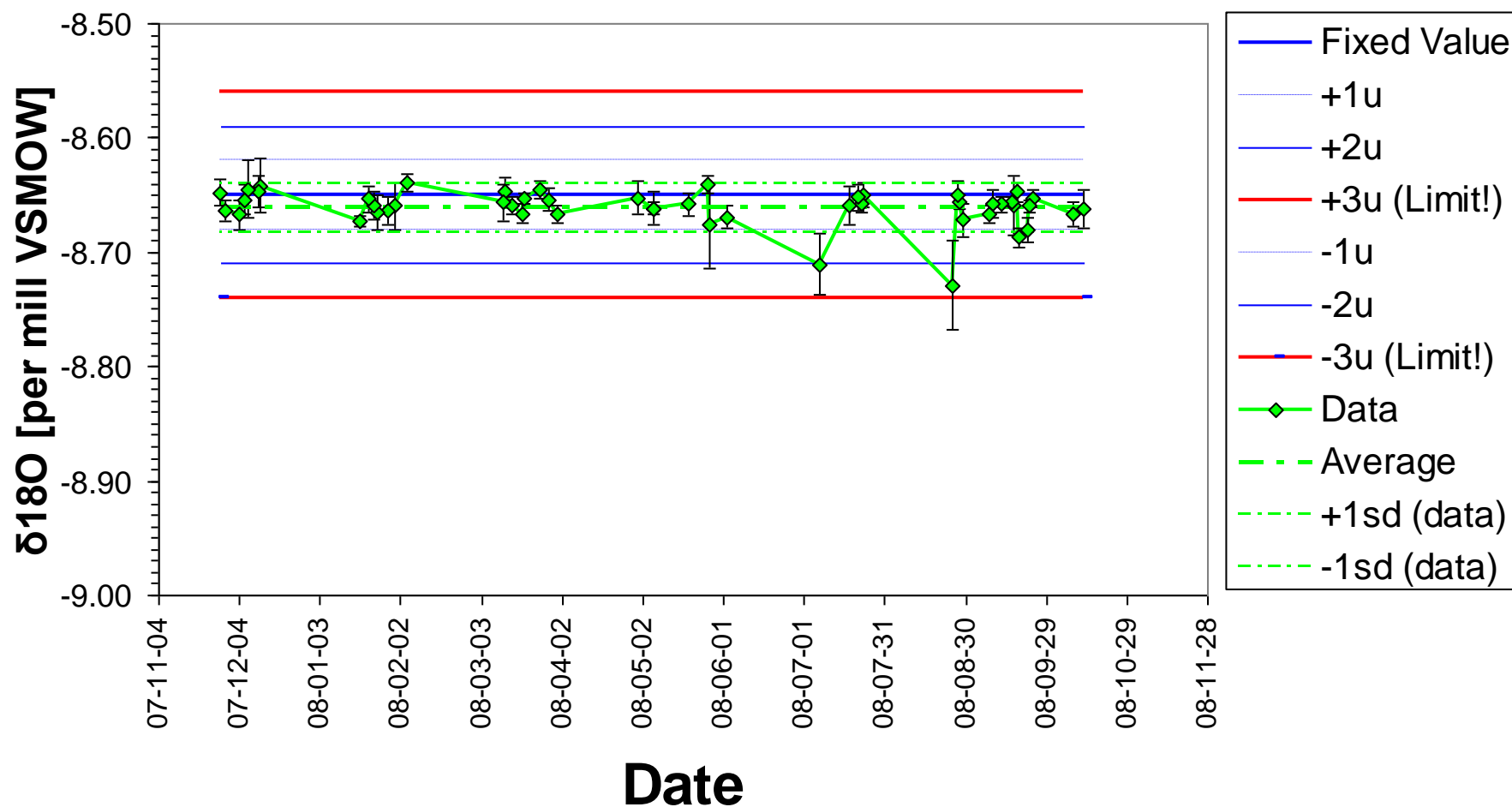
$\delta^{18}\text{O}$ values vs. WS in ‰ (drift corrected data!)

	VSMOW2	SLAP2	xx2	Std6	Std8	GISP/Std9	Std10	Std11
Average	15.307	-41.365	#DIV/0!	6.504	3.778	-9.990	-36.590	15.387
std.dev.	0.028	0.036	#DIV/0!	0.016	0.013	0.009	0.023	0.011
Max	15.34	-41.33	0.00	6.53	3.79	-9.98	-36.57	15.40
Min	15.27	-41.41	0.00	6.49	3.76	-10.00	-36.62	15.37
n	6	5	0	4	5	5	4	6
1	15.34	-41.33		z	3.79	-9.98	-36.62	15.39
2	15.30	-41.40		6.50	3.78	-9.99	-36.60	15.39
3	15.27	-41.34		6.49	3.76	-9.98	-36.57	15.38
4	15.31	-41.35		6.53	3.79	-9.99 rejected		15.39
5	15.34 rejected			6.51	3.77	-10.00	-36.58	15.37
6	15.29	-41.41						15.40
7								



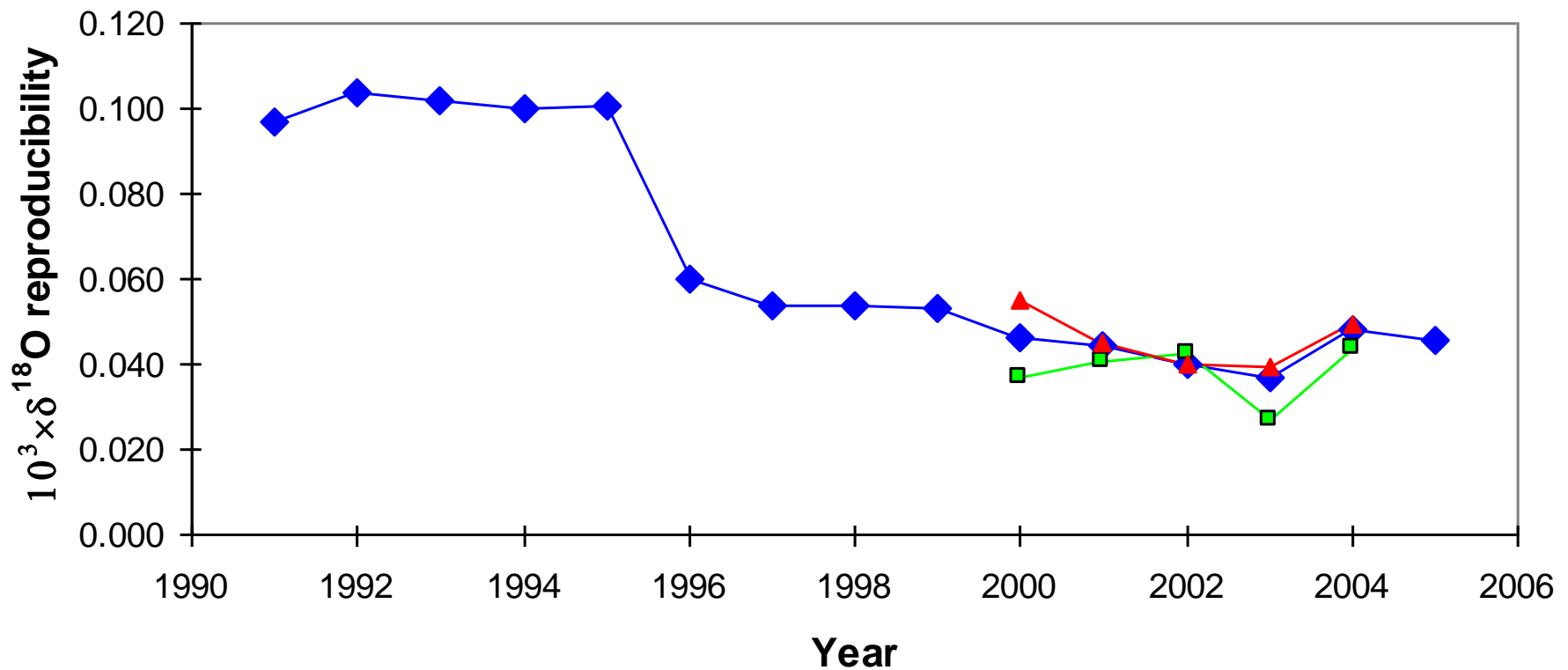
QC-charts

HTW - $\delta^{18}\text{O}$ vs. VSMOW (Delta+) 50 days



Long term reproducibility

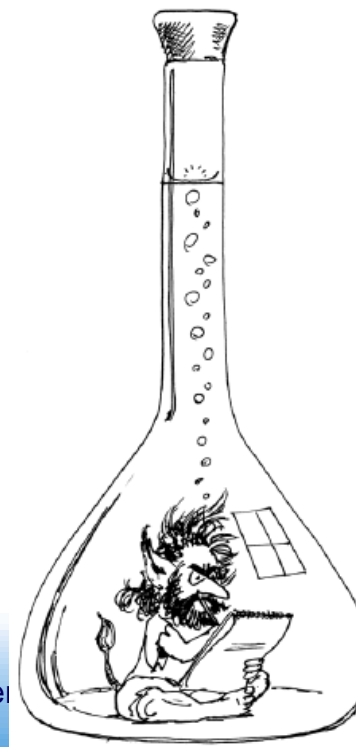
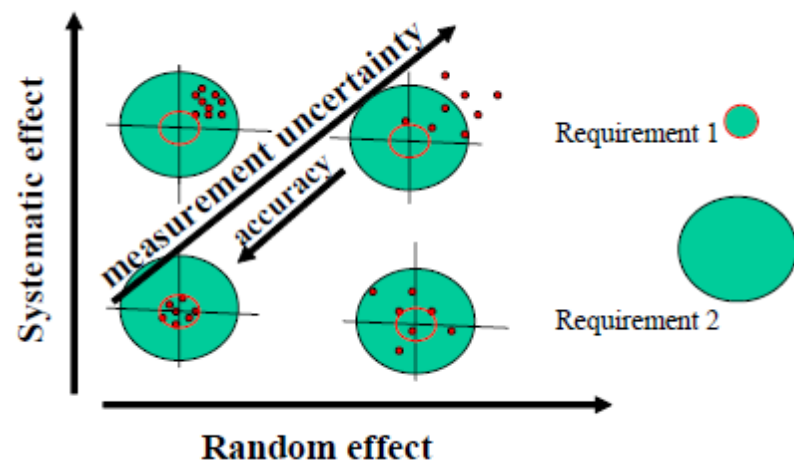
Annual average reproducibility for $\delta^{18}\text{O}$ analyses at IAEA IHL





Internal QUALITY CONTROL

**Handbook for
Chemical Laboratories**



2.f Uncertainty components

- **Repeatability**
- **Reproducibility**
- **Memory**
- **Blanks**
- **Contamination effects**
- **Linearity and drift of measurement system**
- **Calibration of laboratory standards**
- **Measurement of international standards**
- **Bias by used procedures and equipment**

Combined Uncertainty

$$\delta^2 H_{\text{sample}} = \delta^2 H_{\text{cal1}} + (\delta_w^2 H_{\text{sample}} - \delta_w^2 H_{\text{cal1}}) \cdot (\delta^2 H_{\text{cal2}} - \delta^2 H_{\text{cal1}}) / (\delta_w^2 H_{\text{cal2}} - \delta_w^2 H_{\text{cal1}})$$

Combined uncertainty for calibration formula:

$$u(\delta_{\text{sample}}) = \sqrt{\left(\frac{\partial f}{\partial \delta_{\text{cal1}}}\right)^2 \cdot u(\delta_{\text{cal1}})^2 + \left(\frac{\partial f}{\partial \delta_{\text{cal2}}}\right)^2 \cdot u(\delta_{\text{cal2}})^2 + \left(\frac{\partial f}{\partial \delta_{w \text{ cal1}}}\right)^2 \cdot u(\delta_{w \text{ cal1}})^2 + \left(\frac{\partial f}{\partial \delta_{w \text{ cal2}}}\right)^2 \cdot u(\delta_{w \text{ cal2}})^2 + \left(\frac{\partial f}{\partial \delta_{w \text{ sample}}}\right)^2 \cdot u(\delta_{w \text{ sample}})^2}$$

Uncertainty components

Example for the third term only:

Variance from uncertainty (here standard error of the mean ‘sem’) of measured values of calibration standard cal1 at the measurement day:

$$\left(\frac{\partial f}{\partial \delta_{w \text{ cal1}}} \right)^2 \cdot u(\delta_{w \text{ cal1}})^2$$

Corresponding sensitivity factor:

$$\left(\frac{\partial f}{\partial \delta_{w \text{ cal1}}} \right) = (\delta_{\text{cal2}} - \delta_{\text{cal1}}) \cdot (\delta_{w \text{ sample}} - \delta_{w \text{ cal1}}) / (\delta_{w \text{ cal2}} - \delta_{w \text{ cal1}})^2 - (\delta_{\text{cal2}} - \delta_{\text{cal1}}) / (\delta_{w \text{ cal2}} - \delta_{w \text{ cal1}})$$

I THOUGHT I WAS
INTERESTED IN UNCERTAINTY
BUT NOW I'M NOT SO SURE



Don't panic

JOHNSON

Water stable isotope measurements and their associated uncertainty

Main uncertainty components for any measurement on water:

	$\delta^{18}\text{O}$	$\delta^2\text{H}$
Sample measurement repeatability:	$\pm 0.01 \text{ ‰}$	$\pm 0.1 \text{ ‰}$
Sample measurement reproducibility:	$\pm 0.048 \text{ ‰}$	$\pm 0.76 \text{ ‰}$
Daily lab standard measurement:	$\pm 0.03 \text{ ‰}$	$\pm 0.6 \text{ ‰}$
Daily control sample measurement:	$\pm 0.018 \text{ ‰}$	$\pm 0.3 \text{ ‰}$
Linearity correction offset:	$\pm 0.015 \text{ ‰}$	$\pm 0.3 \text{ ‰}$
Measurement of internat. standards:	$\pm 0.02 \text{ ‰}$	$\pm 0.3 \text{ ‰}$
Uncertainty of internat. standards:	$\pm 0.02 \text{ ‰}$	$\pm 0.3 \text{ ‰}$
Combined standard uncertainty:	0.07 ‰	1.2 ‰



3. Implementation tool: SICalib

- Tool for calibration of stable isotope data (hydrogen and oxygen):
- SICalib.xls, an Excel spreadsheet based evaluation tool, developed at the IAEA to deal with calibration and uncertainty issues for the variety of methods for water $\delta^2\text{H}$ and $\delta^{18}\text{O}$ (Gröning 2011, RCM)
- Ideally the same algorithms should be used on all raw data produced by the analytical systems to compile, correct and calibrate
- The same is true for calculation of combined uncertainties

Select the analytical system

Microsoft Excel - SICalib.xls

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Target =

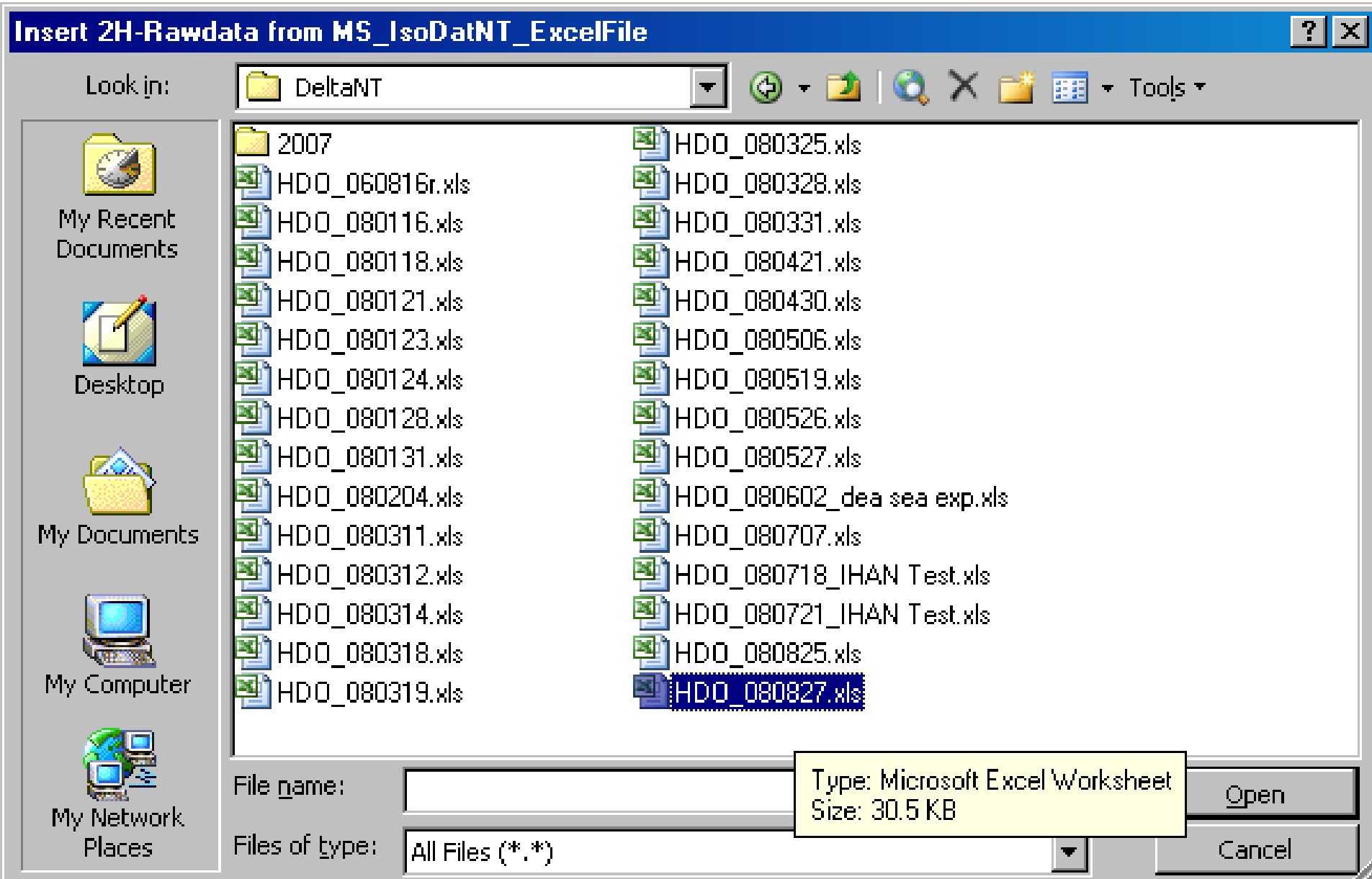
	A	B	C	D	E	F	G	H	I	J	K	L	M
1	SICalib Version 2.10 Manfred Gröning, 2007-12-12												
2	Calibration Data IHL (status as of 2003-12-03)												
3	Selected Calibration Procedure:				Name				delta2H	u(d2H)	delta18O	u(d18O)	Instrument
4					VSMOW				0	0	0	0	DELTA+
5					SLAP				-428	0	-55.50	0	MAT250
6	Delta+: H&O				VSMOW2				0	0.3	0	0.02	LASER
7					GISP				-189.5	0.4	-24.78	0.04	
8	Delta+: H				SLAP2				-427.5	0.3	-55.50	0.02	
9					Std6				-61.1	0.3	-8.70	0.06	
10	Delta+: O				Std7				-4.1	1.2	-0.07	0.03	
11					Std8				-78.4	1.0	-11.29	0.04	
12	Laser: H&O				Std9				-189.1	0.9	-24.77	0.03	
13					Std10				-398.1	0.9	-50.94	0.05	
14	MAT250: H				Std11				-1.4	1.9	0.08	0.05	
15					Std12				-86.9	1.0	-12.01	0.03	
16	IsoPrime: H				OH-1				-3.9		-0.05		
17					OH-2				-30.8		-3.28		
18	IsoPrime: O				OH-3				-61.3		-8.65		
19					OH-4				-109.4		-15.28		
20													
21													
22													
23													
24													

SI_Calib

Draw AutoShapes

Ready

Import the raw data file



Memory correction

Microsoft Excel - fgh080522-1.xls [Read-Only]										
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Type a question for help										
U4										
V	W	X	Y	Z	AA	AB	AC	AD	AE	
1	CalibrationParameters									
2	LS1:	LS2:								
3	std7	std9	std6							Sample
4	300.1	62.5	226.5							Mean:
5	0.9	1.8	1.5							Stddev
6	-4.1	-189.1	-61.1							
7	1.2	0.9	0.3							
8		1.283983								
9	\$AG\$43	\$AH\$43	\$AF\$43							
10	MemoryCorrectionFactor dete			DriftCorrectionParameters			Slope: linearDrift			
11	0.05	used value		used:	-0.0128	0.0000	-0.0049	Update		
12					ConstSlop	Slope(_Slope)	Slope(_Inte	DriftPara		
13	0.055833				-0.0128	0.0000	-0.0049			
14	0.017553				Samples_t	MaxNo.MultAna	TotalCount	TotalSum:		
15					77	32	392	77420		
16	x 3.363766	std6_std7		SampleNa	Slope	Intercept	Count	SumNo		
17	0.047	std7_std9		std6	-0.0133	229.09	32	6000		
18	0.042	std9_66476		std7	-0.0175	303.59	32	6128		
19	x .1330396	66476_67368		std9	-0.0077	63.68	32	6256		
20	x-.225714	67368_67375		66476	no(<7)-1.028811	15747223	4			
21	x-6.318395	67375_67564		67368	no(<7)-.90407226	3646958	4			
22	x-2.480655	67564_67655		67375	no(<7) 9.16413855	047722	4			
23	x-1.074804	67655_66388		67564	no(<7) 1.09002954	4401731	4			

[illegible]

2-Std Calibration & uncertainty eval

Microsoft Excel - SIRawdata_template1

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	Name	Weighted Mean	Combined Mean	UncTypeA	UncTypeB	Mean	Stddev	ESE	count	SS
12	Final summary data for all measured samples									
15	sir10	-56.84	0.07	0.06	0.04	-56.84	0.20	0.04	24	
16	sir8	-23.51	0.04	0.03	0.03	-23.51	0.09	0.02	24	
17	sir3	-9.76	0.05	0.03	0.03	-9.76	0.12	0.02	36	
18	sir1	0.39	0.05	0.04	0.04	0.39	0.15	0.02	36	
19	sir2	-5.29	0.05	0.03	0.04	-5.29	0.13	0.02	30	
20	18o-enr_dil1	175.59	0.31	0.23	0.20	175.59	0.73	0.15	24	
21	18o-enr_dil2	89.02	0.19	0.15	0.12	89.02	0.53	0.11	24	
22	18o-enr_dil3	79.01	0.16	0.12	0.11	79.01	0.40	0.08	24	
23	18o-enr_dil11	-0.74	0.05	0.04	0.04	-0.74	0.11	0.02	24	
24	18o-enr_dil21	-0.86	0.05	0.03	0.04	-0.86	0.06	0.01	24	
25	18o-enr_dil31	-0.90	0.05	0.03	0.04	-0.90	0.09	0.02	24	
26										
27										
28										
29										
30										

ImportedRawdata 2H-Data 2H_Calib 2H-com

Draw [Drawing Tools] AutoShapes [Shape Icons]

Ready [Status Bar Icons]

SiCalib – How can it help you?

- **Requirement: two ILS standards for calibration (to be measured two times per run to assess drift)**
- **Reference values of ILS standards**
- **Knowledge on typical reproducibility for used analytical method (by QC)**
- **Data import possible for all instruments**
- **Consistent memory/drift correction methods**
- **Combined uncertainty automatically calculated**
- **Facilitates proper data reporting**
- **Compatible with LabData – free SQL based database for isotope laboratories (Axel Suckow, CSIRO)**



Thank you for your attention!

More this afternoon...